

Preparation of lead-free ferroelectric BaTi₂O₅ large-sized single crystals by floating zone melting

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The transparent single crystalline BaTi₂O₅ about 6 mm in diameter was successfully grown along *b*-direction by floating zone melting. The effect of foreign element substitution on the dielectric properties of BaTi₂O₅ single crystals was investigated. BaTi₂O₅ is a promising lead-free ferroelectric material used at higher temperatures due to the high permittivity and the high Curie temperature.

Although the phase diagram of BaO-TiO₂ system has been studied for a long time, the ferroelectricity of BaTi₂O₅ (BT₂) has not been known until recently. Our group and Akishige *et al.* independently synthesized single-crystalline BT₂ and reported the significant ferroelectricity only in the *b*-direction. Since BT₂ can be easily decomposed into BaTiO₃ (BT) and Ba₆Ti₁₇O₄₀ (B₆T₁₇) above 1500 K, large-scaled single crystalline BT₂ can be hardly prepared. In order to apply BT₂ for practical applications, the process to prepare *b*-direction oriented BT₂ single crystal in large scale should be developed.

Dried BaCO₃ and TiO₂ powders (purity: 99.99%) were weighed and mixed at the molar ratio of 1 to 2 exactly. A floating zone (FZ) melting apparatus with a xenon heating lamp was used to prepare single crystalline BT₂. The crystal phase was identified by X-ray diffraction (XRD). The dielectric properties were measured in air with an impedance analysis (Hewlett Packard 4194A) from 293 to 1073 K in a frequency (*f*) range between 10² and 10⁷ Hz. A gold paste and wire was used as electrodes.

The as-grown single crystalline BT₂ was transparent and its size was 6 mm in diameter and 30 mm in length. Figure 1 shows the appearance of single crystalline BT₂ prepared by FZ, which was cut perpendicular to the growth direction, *i.e.*, *b*-direction. Ba_{0.99}Sr_{0.01}Ti₂O₅ (BST₂) single crystal 4 mm in diameter and 16 mm in length was also obtained by the FZ. The SrO distribution in the as-grown single crystals prepared by

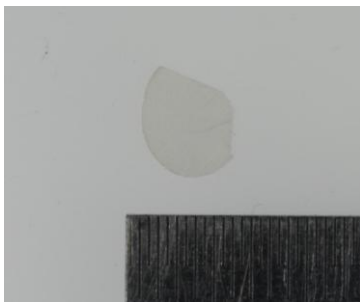


Fig. 1 Appearance of large-sized transparent BT₂ single crystal prepared by FZ.

FZ melting was almost uniform ($\pm 0.02\%$) except the initial and final growth regions.

The peak permittivity (ϵ_{\max}) of single crystalline BST₂ at Curie temperature (T_c) changed depending on SrO content (x). The permittivity of the single crystalline BST₂ showed sharp peaks at the T_c and increased with increasing x until 0.01. Figure 2 demonstrates the effect of x on the T_c and the ϵ_{\max} of single crystalline BST₂. The T_c of single crystalline BST₂ decreased from 748 to 742 K with increasing x from 0 to 0.03. The T_c of BST decreased linearly from 405 to 392 K with increasing x from 0 to 0.03, implying more sensitive to the Sr²⁺ substitution compared to that of BST₂. The solubility limit of Sr²⁺ in BaTi₂O₅ was much lower than that in BaTiO₃ because of the more complicated crystal structure of BaTi₂O₅. The smaller solubility of Sr²⁺ in BaTi₂O₅ may result in the smaller change of T_c than that in BaTiO₃. The maximum permittivity of single crystalline BST₂ had the highest value of 42190 at $x = 0.01$.

The remnant polarization (P_r) and coercive electric field (E_c) of single crystalline BaTi₂O₅ were $1.4 \times 10^{-2} \text{ Cm}^{-2}$ and $0.7 \times 10^6 \text{ Vm}^{-1}$, respectively. The P_r and E_c of the single crystalline BST₂ at $x = 0.03$ were $4.8 \times 10^{-2} \text{ Cm}^{-2}$ and $1.7 \times 10^6 \text{ Vm}^{-1}$, which were much larger than that of single crystalline BT₂, respectively.

Key Words

Lead-free ferroelectric BaTi₂O₅, Single crystal, Floating zone melting

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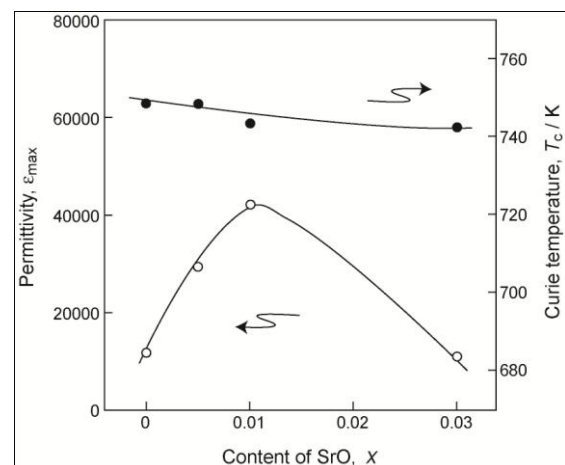


Fig. 1 Effect of SrO content in BST₂ single crystal on the highest permittivity and Curie temperature.