

Highlight of the ICC-IMR Integrated Project on  
“**Mobility and Bonding State in Complex Hydrides**”

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Lithium borohydride  $\text{Li}(\text{BH}_4)$  is one of the complex hydrides [1-4] showing the high-density hydrogen storage property, and also exhibits lithium fast-ion mobility (conductivity) as a novel chemical property. The lithium ion conductivity of  $\text{Li}(\text{BH}_4)$  jumps by three orders of magnitude at approximately 390 K due to its structural transition from the orthorhombic room-temperature (RT) phase to the hexagonal high-temperature (HT) phase [5-6]. Although a wide variety of inorganic lithium ion conductors such as oxides and sulfides have been reported, no lithium fast-ion conductor of hydride except  $\text{Li}(\text{BH}_4)$  has been found since the report of  $\text{Li}_2(\text{NH})$  in 1979. The search and development of the lithium fast-ion conductors is significantly important because they are possibly expected to be applied to solid electrolytes for all solid-state safety lithium ion batteries. From an application point of view, it is highly desired to enhance the lithium ion conductivity of  $\text{Li}(\text{BH}_4)$ -based materials at RT.

During the project period, we have successfully demonstrated the “enhanced conductivities in  $\text{Li}(\text{BH}_4)\text{-LiX}$  ( $X = \text{Cl}, \text{Br}$  and  $\text{I}$ ) [7-9] and  $\text{Li}(\text{BH}_4)\text{-Li}(\text{NH}_2)$  [10]” for the first time. In  $\text{Li}(\text{BH}_4)\text{-LiI}$ , for example, the HT phase of  $\text{Li}(\text{BH}_4)$  can be stabilized even at RT by forming the solid-solutions in the wide composition range, resulting in the lithium ion conductivity ( $4 \times 10^{-5}$  S/cm) more than 3 orders of magnitude higher than that of pure  $\text{LiBH}_4$  ( $2 \times 10^{-8}$  S/cm) at RT. Also in the case of  $\text{Li}(\text{BH}_4)\text{-Li}(\text{NH}_2)$ , two stoichiometric compounds  $\text{Li}_2(\text{BH}_4)(\text{NH}_2)$  and  $\text{Li}_4(\text{BH}_4)(\text{NH}_2)_3$ , show fast-ion conductivities of  $1 \times 10^{-4}$  S/cm at RT due to being provided new occupation sites for lithium ions. The results have been widely announced in the web of IMR: <http://www.imr.tohoku.ac.jp/jpn/result/seika/2009.html#n14>

In addition to the issue on fast-ion mobility, we have intensively studied the “specific surface bonding state in aluminum hydrides  $\text{AlH}_3$  [11-12]”, mainly viewpoint from the high-density hydrogen storage property.

## Selected Publication List (as References for the Highlight)

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